

OFFICE OF NAVAL RESEARCH
END-OF-THE-YEAR REPORT
PUBLICATIONS/PATENTS/PRESENTATIONS/HONORS/STUDENTS REPORT

for

GRANT: N00014-96-1-1056

PR Number 97PR0-6782-00

Polymerized Phthalocyanines for Electrochemical Applications

Peter W. Faguy, Richard P. Baldwin and Robert M. Buchanan

Department of Chemistry, University of Louisville

2320 South Brook St.
Louisville, KY 40292

August 4, 1997

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**OFFICE OF NAVAL RESEARCH
PUBLICATIONS/PATENTS/PRESENTATIONS/HONORS REPORT**

PR Number: 97PR0-6782
Contract/Grant Number: N00014-96-1-1056
Contract/Grant Title: Polymerized Phthalocyanines for Electrochemical Applications
Principal Investigator: Peter W. Faguy
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- a. Number of papers submitted to refereed journals, but not published: 0
- b. Number of papers published in refereed journals (for each, provide a complete citation): 0
- c. Number of books or chapters submitted, but not yet published: 0
- d. Number of books or chapters published (for each, provide a complete citation): 0
- e. Number of printed technical reports/non-refereed papers (for each, provide a complete citation): 0
- f. Number of patents filed: 0
- g. Number of patents granted (for each, provide a complete citation): 0
- h. Number of invited presentations (for each, provide a complete citation): 4
- i. Number of submitted presentations (for each, provide a complete citation): 3
- j. Honors/Awards/Prizes for contract/grant employees (list attached): 0
(This might include Scientific Society Awards/Offices, Selection as Editors, Promotions, Faculty Awards/Offices, etc.)
- k. Total number of Full-time equivalent Graduate Students and Post-Doctoral associates supported during this period, under this PR number: 1.167

Graduate Students: 1.167

Post-Doctoral Associates: 0

including the number of,

Female Graduate Students: 0.417

Female Post-Doctoral Associates: 0

the number of

Minority Graduate Students: 0.5

Minority Post-Doctoral Associates: 0

and, the number of

Asian Graduate Students: 0.667

Asian Post-Doctoral Associates: 0

- l. Other funding (list agency, grant title, amount received this year, total amount, period of performance and a brief statement regarding the relationship of that research to your ONR grant)

ONR - AASERT Award F97 *Polymerized Phthalocyanines for Electrochemical Applications*

h. Invited Presentations

P.W. Faguy, "New Electrocatalysts for Fuel Cell Applications", invited lecture; Department of Chemical Engineering, University of Louisville, Louisville, KY; March 13, 1997.

P.W. Faguy, "Polymerized Tetraaminophthalocyanines: New Electrocatalysts for Fuel Cells", invited lecture; Department of Chemistry and Biochemistry, University of Guelph, Guelph, Ontario, Canada; March 17, 1997.

P.W. Faguy, "Fuel Cells and Batteries in Electric Cars: Research Challenges and Opportunities", invited lecture; Department of Chemistry, West State College of Georgia, Carrollton, GA; April 11, 1997.

J.E. Pavez, J.P. Shaw, F. Bailly, R.P. Baldwin, R.M. Buchanan and P.W. Faguy, "The Electrochemical Polymerization of Tetraaminophthalocyanines to Form Conducting Polymers", invited talk; presented at the Electrochemistry Symposium at the Canadian Society for Chemistry Meeting, Windsor, Canada; June 1-4, 1997.

i. Submitted Presentations

M. Huang, P.W. Faguy, R.P. Baldwin, J. Shaw, J. Pavez, J. Wang, F. Bailly and R.M. Buchanan, "Electrocatalytic Properties of Poly-Co-Tetraaminophthalocyanine Modified Electrodes", presented at the 213th National ACS Meeting, San Francisco, CA; April 13, 1977, paper 286.

J. Pavez, J. Shaw, F. Bailly, R.P. Baldwin, R.M. Buchanan and P.W. Faguy, "Polymerized Phthalocyanines for Electrochemical Applications" presented at the 3rd Annual Kentucky EPSCoR Conference, Lexington, KY; May 5, 1997.

M. Huang, J. Pavez, J. Shaw, F. Bailly, R.P. Baldwin, R.M. Buchanan and P.W. Faguy, "Oxygen Reduction on Poly-Cobalt Tetraaminophthalocyanine-modified Electrodes", presented at the 191st Meeting of the Electrochemical Society, Montreal, Canada; May 8, 1997, Abstract #1096.

k. Graduate Students and Post-doctoral Associates

Hanlin Nie, Ph.D. student; June 1, 1996 to July 31, 1996; 16.7% FY97 FTE.

Min Huang, Ph.D. student; August 1, 1996 to August 31, 1996; 8.3% FY97 FTE.

Jianping Wang, Ph.D. student; August 1, 1996 to December 31, 1996; 41.7 FY97 FTE.

Jorge Pavez, visiting graduate student, University of Santiago, Santiago, Chile.
December 1, 1996 to May 31, 1997, 50% FY97 FTE.

l. Other Funding

ONR - AASERT Award F97 *Polymerized Phthalocyanines for Electrochemical Applications*
\$120,000 cumulative; 06/01/97 - 05/31/00

Where the present grant addresses the application and development of polymerizable phthalocyanines for the electrocatalytic reduction of molecular oxygen for fuel cells, the AASERT grant focuses on the same polymerizable macrocycles with application to electrocatalytic amelioration of environmental toxins and pollutants.

a. Principal Investigator:

Peter W. Faguy

b. Phone Number:(502) 852-6550 (voice) (502) 852-8149 (FAX)
e-mail: pwfagu01@homer.louisville.edu**c. Cognizant ONR Program Officer:**Richard T. Carlin, Electrochemical Science and
Technology, Chemistry and Physics, S&T Division**d. Program Objective**

The main objective of our research program is to develop new electrode materials for oxygen reduction based on polymeric aminophthalocyanines, poly-MTAPc's. Reaching this objective depends on successful outcomes in the following endeavors:

- optimization of monomer syntheses, of polymer formation, and of transition metal exchange in the polymer;
- characterization of O₂ reduction at poly-MTAPC-modified electrodes, in terms of both mechanism and kinetics;
- description, at a molecular level, of the polymer film based on electroanalytical and spectroscopic studies.

e. Significant Results

A new metal-free phthalocyanine, tertakisamino-phthalocyanine (H₂TAPc), has been synthesized and metallated successfully with Co²⁺. The TAPc monomers have been characterized by a variety of electrochemical and spectroscopic techniques. The redox properties of dissolved, adsorbed and polymerized films of H₂TAPc and CoTAPc have been studied in non-aqueous (DMF and DMSO) and aqueous solutions. Both of these monomers electropolymerize on ITO, Pt and GC electrodes in DMF and DMSO at ~+0.8 V vs. Ag/AgCl. Films polymerized from H₂TAPc can be metallated by electrochemical cycling in DMSO (0.1 M TBAP) containing CoCl₂. Poly-H₂TAPc films catalyze hydrogen evolution with onset near -0.8 V in acidic solutions. Poly-CoTAPc modified electrodes reduce molecular oxygen with an onset potential of +0.2 V vs. Ag/AgCl in 0.05 M H₂SO₄ and with no loss of activity over thousands of cycles. O₂ reduction is also electrocatalyzed in basic solution, again with no apparent loss of activity. In preliminary experiments, the films have retain their activity in the presence of methanol and chloride ion.

f. Work Plan 1997 - 1998

With respect our program goals and our 96/97 accomplishments, there are several obvious short-term objectives:

- 1) confirm the proposed kinetics and mechanism for O₂ reduction on poly-CoTAPc-modified electrodes using rotating ring-disc hydrodynamic studies (underway);
- 2) demonstrate generality of electrochemically induced central metal exchange process (Ni²⁺ and Fe²⁺ studies underway);
- 3) quantify the extent of stability for poly-CoTAPc-modified electrodes toward methanol and chloride ion and for long cycle times, under real PEM fuel cell conditions.

Other objectives which will be addressed in the next year (and possibly extend into the following year), which involve more complicated experimental design and procedures include:

- 4) establish the mechanism of electrochemical polymer film formation using electrochemical methods and hybrid *in situ* techniques such as FTIR, Raman and QCM;
- 5) optimize film formation with respect to electrode performance and stability and in terms of fabrication methodology (e.g.; polymerization conditions: cycling / potentiostatic /galvanostatic, anion composition, solvent, etc.);
- 6) initiate studies into poly-FeTAPc-modified electrodes for oxygen reduction.

g. Current Graduate Students

Francois Bailly: M.Sc. expected September, 1998.

Pablo Condes: Ph.D. expected 2001.

Jorge Pavez: Ph.D. (University of Santiago), leaving project December, 1997.

William Todd: pre-M.Sc. student, leaving project December, 1997.

Polymerized Phthalocyanines for Electrochemical Applications

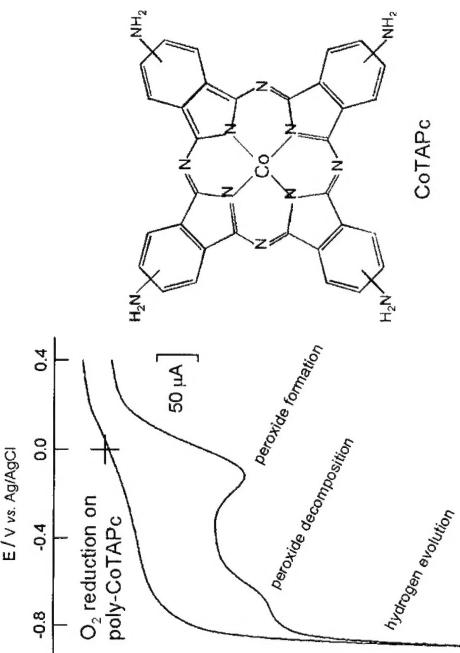
Faguy/Baldwin/Buchanan, University of Louisville

Technology Issues: Fuel cells which employ transition metal macrocycles as O₂ electrocatalysts demonstrate less than optimal performance due to loss of activity and/or poisoning; new electrode materials are needed.

Objective: Develop new polymeric macrocyclic electrocatalysts which demonstrate a 4 e⁻ reduction pathway, low overpotentials and long-term stability.

Approach:

- Modify existing synthetic routes to prepare various tetraaminophthalocyanines.
- Develop novel metallation methods to make MTAPc's.
- Optimize the electro-oxidation process leading to polymer film formation.
- Study O₂ reduction kinetics at poly-MTAPc-modified electrodes.
- Characterize the polymer films using electroanalytical and spectroscopic methods.



Accomplishments:

- Demonstrated a novel route to poly-CoTAPc films via electrochemical exchange of Co²⁺ in the poly-H₂TAPc film.
- Achieved facile O₂ reduction at poly-MTAPc films:
 $M=H_2 \rightarrow 2 e^-$
 $M=Mg \rightarrow 2 e^-$
 $M=Co \rightarrow 4 e^- (2+2)$
- Demonstrated long-term stability for poly-CoTAPc films as O₂ reduction catalysts.
- Exciting possibilities for transition metal and non-transition metal hybrid poly-M, M'TAPc films for electrocatalysis and electroanalysis now exist.

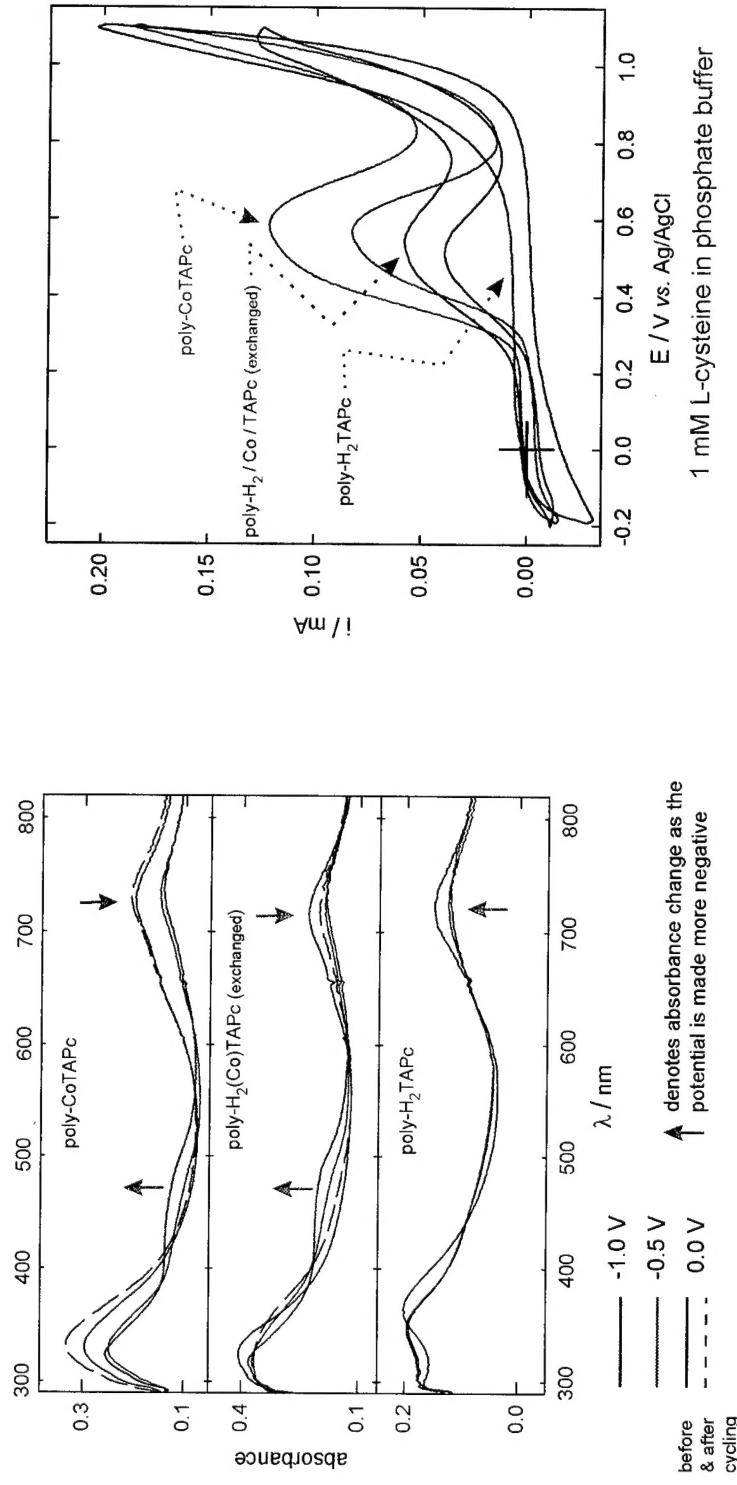
Impact:

- New poly-TAPc-based O₂ reduction catalysts, demonstrating stability and performance, predict a new chapter for research into macrocyclic fuel cell catalysts.
- Exciting possibilities for transition metal and non-transition metal hybrid poly-M, M'TAPc films for electrocatalysis and electroanalysis now exist.

Polymerized Phthalocyanines for Electrochemical Applications
Department of Chemistry, University of Louisville

ONR EOY 96/97
Report Part III-2

electrochemically induced metallation of poly-H₂TAPc films

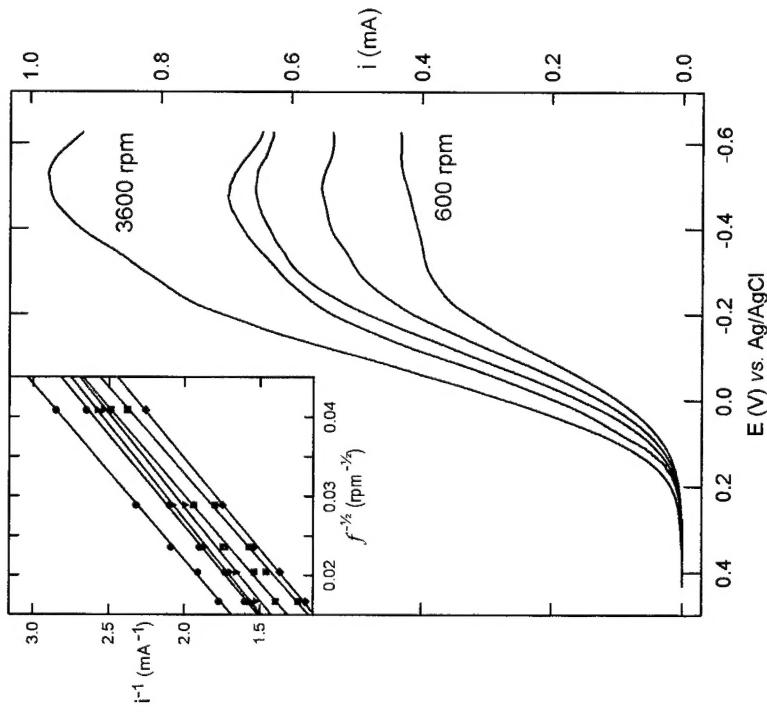


Accomplishment & Impact: No previous demonstration of an electrochemically induced metallation for a macrocycle-modified electrode exists. UV-visible spectroelectrochemistry (LH panel) and electrocatalysis (RH panel) confirm that Co²⁺ is being incorporated as the central cation in the TAPc macrocycle. From robust, easily formed poly-H₂TAPc films, a wide variety of transition metal and mixed metal electrocatalysts could be made. Regeneration of active transition metal centers could be possible for fuel cell electrodes, and for electroanalytical and electrochemical remediation applications.

Polymerized Phthalocyanines for Electrochemical Applications
Department of Chemistry, University of Louisville

ONR EOY 96/97
Report Part III-3

4e⁻ reduction of O₂ on poly-CoTAPc films



for $-220 \text{ mV} > E > -380 \text{ mV}$ (higher η region):

► all lines parallel (r^2 values > 0.990) → reaction is 1st order with respect to O₂

► slope⁻¹ = β (Levich parameter) → $0.021 < \beta < 0.028$
 $\bar{\beta} = 0.023 \text{ mA}/(\text{rpm})^{1/2}$ 4e⁻ process

Accomplishments & Impact: The work to date has indicated that poly-CoTAPc films electrocatalyze the four electron reduction of oxygen to water in acidic media and to OH⁻ in basic media. The mechanism most probably involves two sequential 2e⁻ reductions. The onset potential for O₂ reduction, the behavior at high rotation rates, and the robustness of the film towards electrochemical cycling all reflect favorably on the potential of these novel polymeric films for fuel cell applications. The film stability is present in acidic and basic solutions. Furthermore, oxygen reduction on poly-CoTAPc-modified electrodes is not adversely affected by the presence of either methanol or chloride ion in the electrolyte solution.

O₂ reduction RDE study in 0.05 M H₂SO₄